

**EXTRUDED POLY(ETHYLENE–CO–VINYL ALCOHOL) COMPOSITE FILMS  
REINFORCED WITH CELLULOSIC FIBERS ISOLATED FROM TWO LOCAL  
ABUNDONATES PLANTS**

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**ABSTRACT**

El Diss and El Retma fibers are in abundance in North Africa, collected from Setif (Algeria) and have been treated to isolate cellulose fibers with toluene-ethanol and HNO<sub>3</sub> to improve their dispersion into EVOH matrix. SEM micrographs and FTIR analyses of the treated fibers confirmed the elimination of non cellulose materials and their crystallinity was estimated by DRX. Thermal analyses by TGA indicate a slight improvement compared to the raw fibers.

Composites were also prepared by incorporation of the cellulose fibers in EVOH matrix. FTIR results and water absorption behavior indicate a reaction between the treated fiber and EVOH matrix by forming hydrogen bonds. Thermal properties of the composites reported by DSC results decreased compared to neat EVOH. The addition of cellulose fibers led to an increase in the loss and storage modulus and melt viscosity of the composites.

**Keywords:** El Diss fibers, El Retma fibers, cellulose, biocomposites

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## 1. INTRODUCTION

Environmental pollution from consumed plastics becomes serious because of their nonbiodegradability. Thus, there has been growing interest in the developing of biobased products to replace nonbiodegradable plastics and reduce the dependence on fossil fuel [1,2].

Biopolymers presented three main materials: The first is synthetic polymers with characteristics groups susceptible to hydrolysis attack by microbes, the second is composed of naturally occurring processible bacterial polymers as like as lignocellulosic fibers [1,3,4]. The third class is blends of polymers and additives that are readily consumed by microorganisms [5]. Cellulose based composites offers many advantages such as a high specific strength and stiffness, a low hardness which reduce wear on processing equipment, a relatively low density, a low cost and biodegradability. In fact, composites reinforced by cellulosic fibers may eventually be recycled or burned to recover heat, without the production of residues or toxic products [6,7].

Natural fibers also known as cellulosic or lignocellulosic fibers are principally composed by cellulose, lignin and hemicellulose, biosynthesized by plants, some amoebae, sea animals, bacteria and fungi. Their properties depends in chemical composition, internal fiber structure, microfibril angle, cell dimensions and structural defects, which differ from different parts of a plant as well as from their origin. The mechanical properties of natural fibers also depend on their crystalline organization [8,9].

Composites with lignocellulosics fillers found application in household, ornament and packaging uses, agricultural and biomedical applications, nonstructural or semistructural building products such as decking, fencing, siding, window frames, and roof tiles [7,10,11]. However, the inclusion of cellulose fibers as fillers for polyolefins has some drawbacks. The incompatibility, the poor interfacial adhesion of polar cellulose and nonpolar matrix and the poor fiber dispersion leads to poor mechanical properties [12].

Many efforts have been conducted to create interaction between the thermoplastic matrix and filler [6,13]. Numerous studies have used the functionalization of polyolefins by maleic anhydride (MA), also Isocyanates and silanes are used as coupling agents. Compatibilizers have been investigated which was found to act, not only as compatibilizer but also as impact

modifier. Another way to improve adhesion in thermoplastic/cellulose composites is crosslinking [11,13,14]. Surface cellulosic fillers can be also modified by physical and chemical methods mainly based on the reactivity of the cellulose hydroxyls [11,15]. Such as alkali treatment, bleaching, acetylation and steaming [16-18].

Ethylene-vinyl alcohol (EVOH) is a hydrophilic semicrystalline copolymer with excellent gas-barrier properties to oxygen and organic compounds. It's considered to have the superior gas barrier properties compared to the most of the polymeric materials depending on the copolymerization ratio of ethylene and vinyl alcohol. Thus, EVOH is widely used in food packaging industry, also because EVOH has a high transparency and easy processability on a wide range of conventional coextrusion processing [19,20].

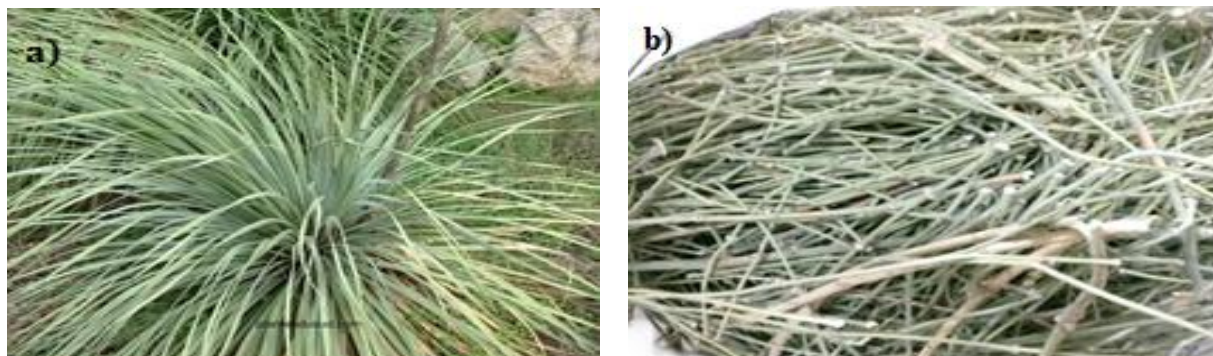
Several blends of vinyl alcohol–ethylene copolymers (EVOHs), and polysaccharides, such as starch, corn fiber and sugar cane bagasse have been reported [21]. Morsyleide and al. [17] studied the incorporation of treated coconut fibers into formulations with starch and EVOH and prived to improve the thermal stability, increased the tensile strength of the composites and also improved values of Young's modulus.

El Diss and El Retma fibers are a wild plants in abundance in the mountains of North Africa and in this case exactly collected from the Medjouness mountains of Ain El Kebira (Setif), Algeria. This present study was carried out to investigate the effects of the chemical treatments of two types of fibers after incorporation in ethylene vinyl alcohol on their thermal, morphology, and tensile properties.

## **2. EXPERIMENTAL**

### **2.1. Materials**

The raw fibers used in this work are El Diss and El Retma, and were come from the region of Setif.



**Fig.1.** El Diss plant (a), El Retma plant (b)

Reagents that were used to extraction were Ethanol, Toluene which are from BIOCHEM chemical pharma and Nitric acid ( $\text{HNO}_3$ ) is purchased from Sigma Aldrich.

EVOH/Cellulose formulations were based on EVOH (44 % ethylene) provided by Sigma Aldrich with the following physical characteristics: pellets density: 1.140, melting point:  $165^\circ\text{C}$ .

## 2.2. Fibers treatments

The aim to this chemical treatments was the elimination of hemicellulose and lignin for the extraction and isolation of cellulosics fibers.

El Diss and El Retma cellulosic fibers was prepared by using the method as following:

The raw fibers were washed, dried and then cuted, powdered in a coffee grinder and sieved ( $100\ \mu\text{m}$ ) to uniformize the particle sizes. The powder was then treated with 50:50 (v:v) toluene-ethanol mixture for 4h in a flask equipped with condenser. After several washes with distilled water, the precipitate was dried at  $105^\circ\text{C}$  for 24h. A treatment was then carried out 3 times with a mix of  $\text{HNO}_3$  + ethanol solution in the propotion (1/4). The solution is carried out to ebulation in a water bath for 1 h under magnetic stirring and condenser. A successif washes was carried out with hot water until complete elimination of the solution. The obtained powder was dried at  $105^\circ\text{C}$  until a constant weight. The final product is a yellow powder.

## 2.3. Composites preparation

The cellulose fibers and EVOH were previously placed in an oven set at  $50^\circ\text{C}$  overnight to eliminate an eventual adsorbed water on the surface. Then, the two treated fibers (El Diss  $\text{CF}_D$

and El Retma CF<sub>R</sub>) were mixed respectively with EVOH into a co-rotating twin-screw configuration (MiniLab microcompounder by Thermo Scientific Haake, Germany), at 200°C, screw speeds at 100 rpm and a time of recirculation designed for 8 min. Two formulations were prepared with two different rate charge (5 and 10%). All samples were prepared with the same procedure, the EVOH/cellulose fibers mix was added in three aliquots (EVOH, cellulose fibers then EVOH) because of the high volume of cellulose, which could cause the blockage of the machine.

## 2.4. Characterization methods

### 2.4.1. Fiber's characterization methods

#### a. FT-IR spectroscopy

Chemical composition study of the fibers after treatment was conducted through FTIR spectroscopy. A FTIR VERTEX 70v was exploited to provide the spectrum of each sample. Spectra were taken at a resolution of 4 cm<sup>-1</sup> and in the range of 4000–400 cm<sup>-1</sup> in the absorbance mode.

#### b. X-ray diffraction analysis (XRD)

The degree of crystallinity and the treatment effect on the morphological crystallinity of the samples was estimated by XRD analysis on D8 Advance system (Bruker) apparatus using CuK<sub>α</sub> radiation in the range  $2\theta = 2-70^\circ$  with a step of 0.01/S. Crystallinity index of the raw fibers were obtained using the following equation:

$$Ci = (I_{002} - I_{amorph}) / I_{002} \times \% \quad (1)$$

Where  $I_{002}$  is the maximum intensity of the 002 peak which represents both crystalline and amorphous material and  $I_{amorph}$  is the lowest height which represents amorphous material only.

#### c. Thermogravimetric analysis (TGA)

The thermal behavior of the treated and untreated fibers were determined in a temperature range of 20–700 °C, at a heating rate of 10 °C/min, in a nitrogen environment (Purge rate details : Balance chamber flow rate = 30 cm<sup>3</sup>/min, furnace flow rate = 150 cm<sup>3</sup>/minute) using a Mettler Toledo TGA/DSC.

#### d. Scanning electronic microscopy (SEM)

The morphological characterization of the fibers samples were done with scanning electronic microscopy (SEM) using JSM-6460 apparatus.

#### **2.4.2. Characterization methods of EVOH/treated fibers composites**

##### **a. FT-IR spectroscopy**

Molecular interactions between matrix and charges were established by FTIR. Analysis of the EVOH/CF composites films were performed using a VERTEX 70v spectrometer in the absorption mode in the range of 4000-400  $\text{cm}^{-1}$  at a nominal resolution of 4  $\text{cm}^{-1}$ .

##### **b. Differential scanning calorimetry (DSC)**

Thermals characteristics were determined by DSC and were performed using a Mettler Toledo DSC 3 calorimeter under nitrogen flow and heating ramp of 10°C/min. The following heating program has been carried out : from 25°C to 200°C, 2 min at 200°C, from 200° to 0°C then 2 min at 0°C, finally from 0°C to 200°C. The crystallinity degree (%  $\text{c}$ ) for each blend was calculated according to eq. (2):

$$\% \text{ c} = 100 \frac{H_m}{(W H_m^\circ)}$$
 (2)

Where  $H_m$  is the melting enthalpy of the blend sample (J/g),  $H_m^\circ$  is the melting enthalpy (169.2 J/g) of the EVOH sample assuming 100% crystallinity as in pure polyvinyl alcohol (PVOH), and W is the EVOH mass fraction in the composite [22].

##### **c. Rheometer**

An Anton Paar CTD 450 Physica MCR 301 rheometer, which works in parallel-plate oscillatory mode, was used to study the rheological behavior of materials at the temperature of 200 °C. Before characterizations, specimens were dried under vacuum and compression molded to the thickness of about 1 mm. Linear viscoelastic region was determined using dynamic strain sweep tests in the range of 0.01–10% strain at the constant frequency of 1 Hz. A strain of 5.0%, corresponding to the linear viscoelastic domain, was chosen to perform dynamic measurements over a frequency range of 0.01 to 20 Hz. For each material, three tests were made to verify the reproducibility of the results.

##### **d. Water absorption measurements**

Films were cut into samples of approximately 1\*1 cm and were dried in a vacuum oven at 100°C for 24h. Weight of film samples were measured using an analytical balance with a

resolution of 0.0001g (Metler Toledo). The samples were then immersed into a closed vessel containing water at room temperature for 72 h, and water uptake measurements were recorded at 24-h intervals.

$$\text{Water uptake (\%)} = (M_t - M_0)/M_0 \times 100 \quad (3)$$

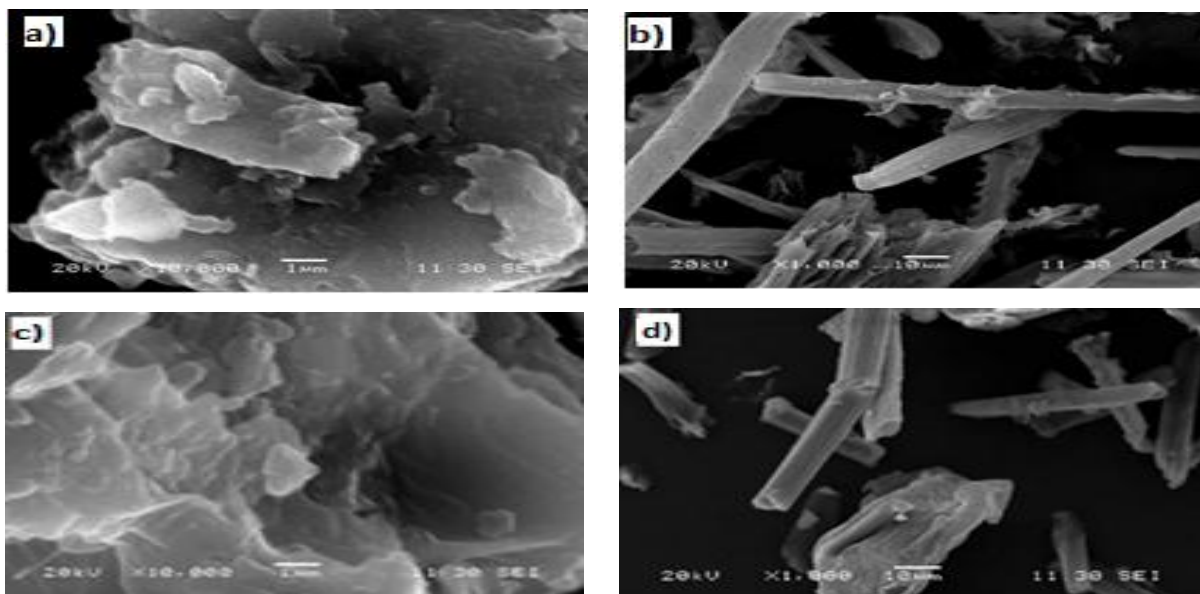
Where  $M_t$  is the mass of the sample at time  $t$  and  $M_0$  is the mass of the sample before insertion into the water [23].

### 3. RESULTS AND DISCUSSION

#### 3.1. Fibers characterization

##### a. Scanning electronic microscopy (SEM)

The modifications occurred to the raw fibers after treatments were investigated by SEM and presented in figure 2. The SEM images (2a and 2b, 2c and 2d) showed the morphologies of untreated and treated fibers of El Diss and El Retma respectively.



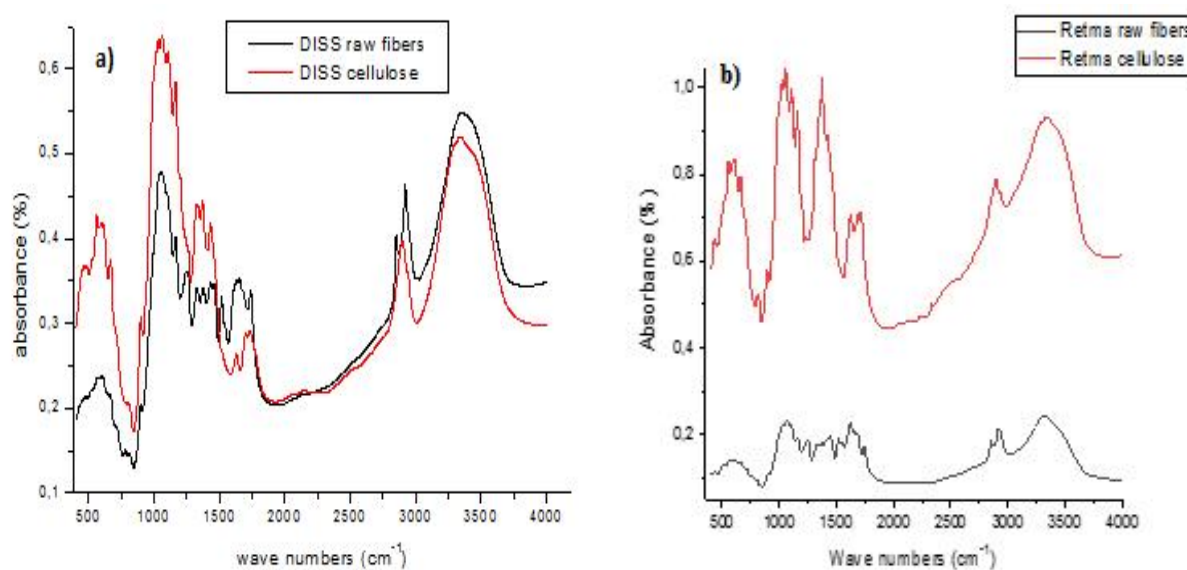
**Fig.2.** SEM images of El Diss raw fibers (a), El Diss treated fibers (b), El Retma raw fibers (c), El Retma treated fibers (d)

The SEM images for the two untreated fibers (El Diss, El Retma) exposed entities with different shapes and sizes with a diameter between 16 - 85  $\mu\text{m}$  for El Diss fibers and 16 - 66  $\mu\text{m}$  for El Retma ones. After treatments, the SEM images showed long fibers, well separated

from each other and the diameter of the two fibers decreased and varied between 4  $\mu\text{m}$  and 10  $\mu\text{m}$  for El Diss fibers and between 5  $\mu\text{m}$  and 9  $\mu\text{m}$  for El Retma ones. Significant morphological changes can be also observed due to the loss of mass after removing extracellular compounds as lignin and hemicellulose to isolate cellulose. The cellulose samples showed a smoother surface compared to the initial fibers [24].

### b. Fourier transform infrared spectroscopy (FTIR)

The chemical structure of the El Diss and El Retma fibers before and after treatments were studied by FTIR analysis and the results are reported in figure 3.



**Fig.3.** FTIR spectra of El Diss (a) and El Retma (b) raw fibers and their celluloses

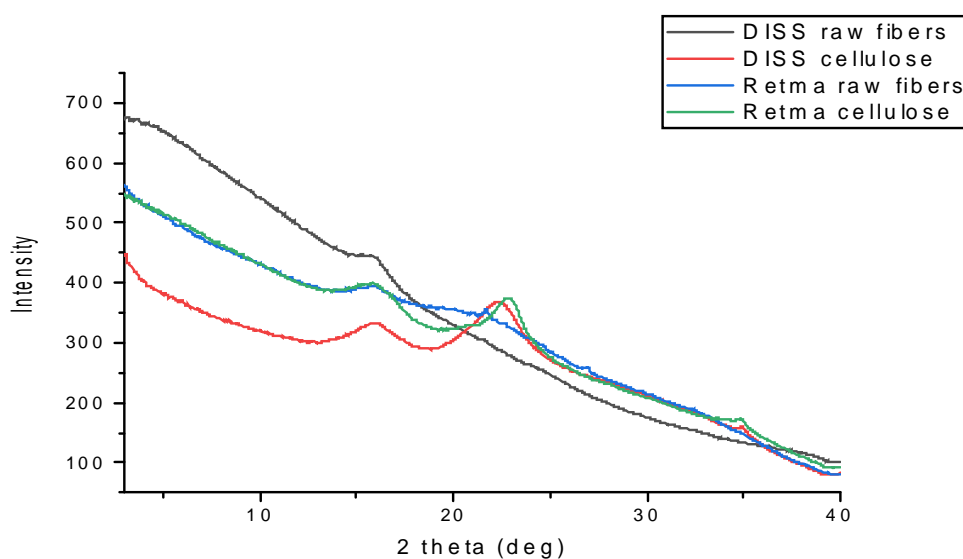
The FTIR spectra corresponding to the two fibers present two main bands at 3300 and 684  $\text{cm}^{-1}$  which were attributed to stretching vibrations and the out-of-phase bending vibrations respectively of O-H in cellulose and lignin, the second band is at 2875  $\text{cm}^{-1}$  which presents the stretching vibrations of the aliphatic C-H of lignin. The absorbance peaks at 1730 and 1427  $\text{cm}^{-1}$  were attributed to stretching vibrations and bending of the hydrogen-carbonyl band in hemicellulose and  $\text{CH}_2$ , respectively. Lignin structure presents also a band at 1500-1600  $\text{cm}^{-1}$  corresponding to C=C bonds in aromatic rings. The hydrophilic character of the cellulosic fibers highlights a band of adsorbed water around 1630  $\text{cm}^{-1}$  [25-27]. The peaks around 1110 and 1060  $\text{cm}^{-1}$  correspond to skeletal vibrations of C-O stretching and the peak at 1161  $\text{cm}^{-1}$



is due to asymmetric stretching of C-O-C of glucose ring in cellulose structure [28].

After fibers treatments, the elimination of extracellular substances was confirmed by the absence of the lignin bands at  $2875\text{ cm}^{-1}$  and  $1500\text{ cm}^{-1}$  in the spectra of the treated fibers [29]. Also, the characteristic bands of hemicellulose and pectin disappear from El Diss and El Retma celluloses spectra.

### c. X-ray diffraction analysis (XRD)



**Fig.4.** X-ray diffraction patterns of El Diss, El Retma raw fibers and their cellulose

Figure 4 displays the crystallinity of the treated and untreated El Diss and El Retma fibers by X-ray diffraction analysis.

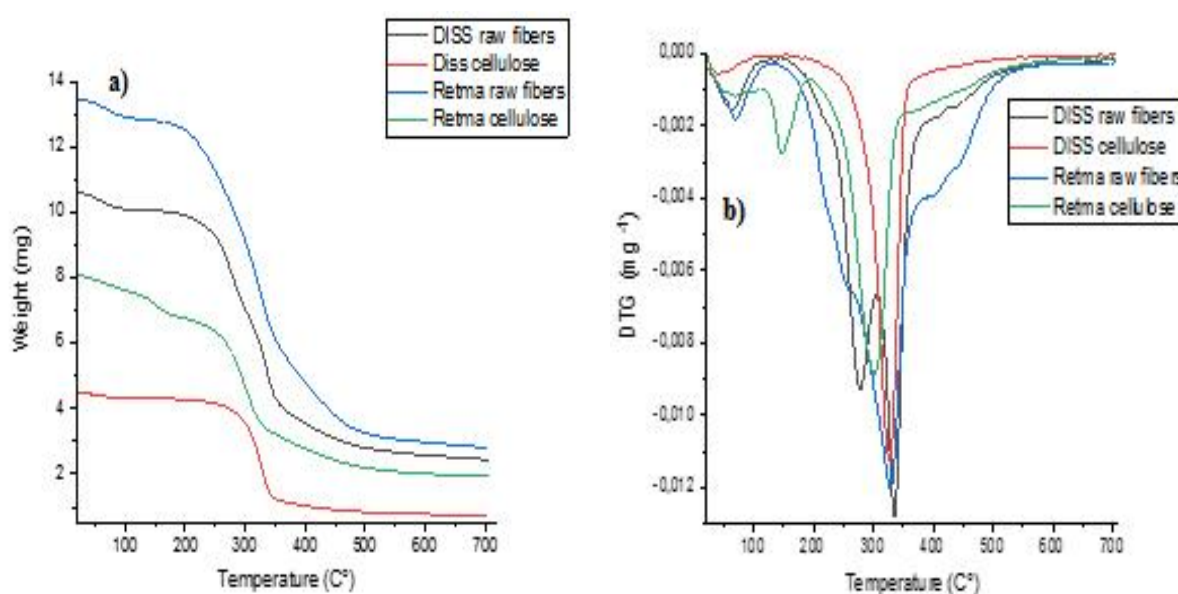
The raw and cellulose fibers of El Diss and El Retma have a typical XRD patterns of a semi-crystalline material constituted of an amorphous and crystalline parts [30]. Native cellulose presents a diffraction signals at  $2\theta$  values of  $15^\circ$ ,  $16^\circ$ ,  $22,5^\circ$  and  $34^\circ$ , attributed to the diffraction planes 101, 101, 002 and 040, respectively [29,31,32].

The raw fibers display only one peak at  $2\theta$  of  $15.7^\circ$  assigned to the (101) typical reflection plane of cellulose I, unlike the treated fibers which exposed a second peak at about  $22.4^\circ$  for El Diss cellulose and at  $22.9^\circ$  for El Retma one, attributed to the (002) reflection plane of cellulose I [33], because of the large peak attributed to the important part of amorphous substance in raw fibers which prevents the appearance of the second peak. Also, the peaks

exhibited at  $15.7^\circ$  are more intense in the cellulose patterns than in the raw fibers ones indicating an increase in the crystallinity after treatments. The determination of the crystallinity using equation 1 gives values of 21.19 % for El Diss cellulose fibers, while for El Retma cellulose fibers it approximates 13.94 %.

#### d. Thermogravimetric analysis (TGA)

The TGA thermograms of the raw fibers of El Diss, El Retma and their cellulose are obtained and showed in figure 5. The analysis were carried out to highlight the influence of the treatments on the thermal degradation characteristics of the fibers.



**Fig.5.** TG (a) and DTG (b) curves of El Diss and El Retma raw fibers and their cellulose

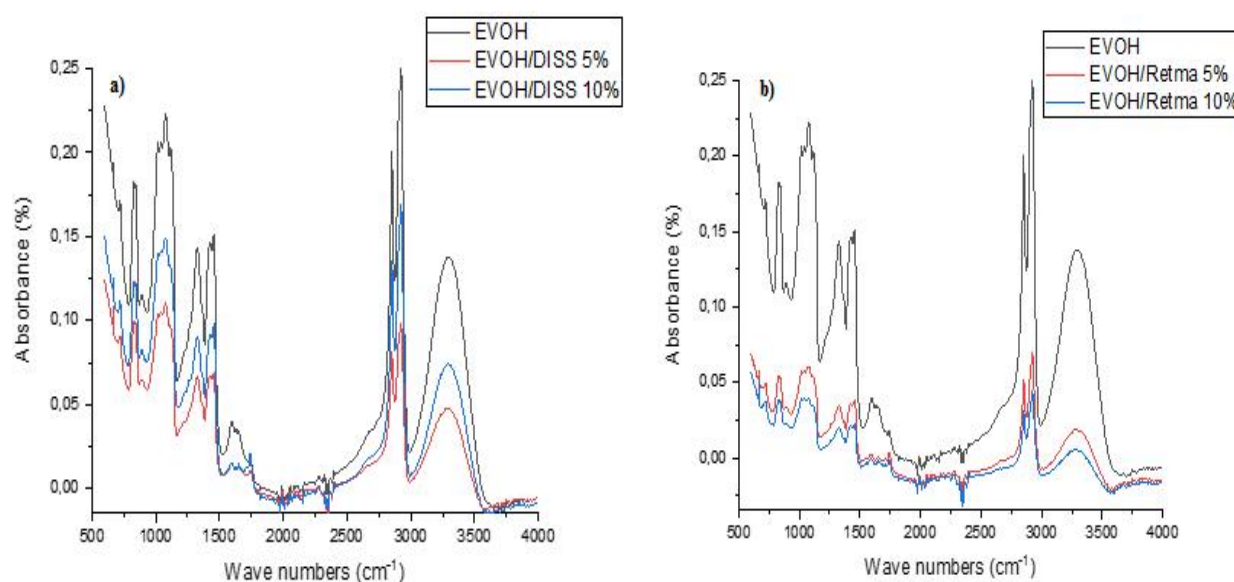
The thermal degradation of the raw fibers of both El Diss and El Retma plants exposed three main weight loss regions indicating the presence of different components which decomposed at different temperatures (hemicellulose, lignin and cellulose). The first loss is mainly due to moisture evaporation of the adsorbed water, the second region temperature occurring between 220 and 350°C is mainly attributed to the thermal depolymerization of hemicelluloses and the cleavage of the glycoside bonds of cellulose. The peak that appears above 350°C is attributed to lignin decomposition, it has low degradation rate and decompose at lower temperature than cellulose and it has more heat resistant than hemicelluloses [34,35].

The thermograms of the treated fibers display two stages decomposition process; the first

one corresponds to moisture evaporation and the second to the cellulose decomposition. The peaks corresponding to the paracellulosic components disappeared confirming their elimination by the different fiber treatments in the case of El Diss fibers but still exist in the El Retma thermograms suggesting that these components have not been totally removed and the parameters treatments are not efficient for El Retma fibers. It is observed that the cellulose of the two plants degraded at a slight higher temperatures compared to the raw fibers.

### 3.2. Composites characterization

#### a. Fourier transform infrared spectroscopy (FTIR)



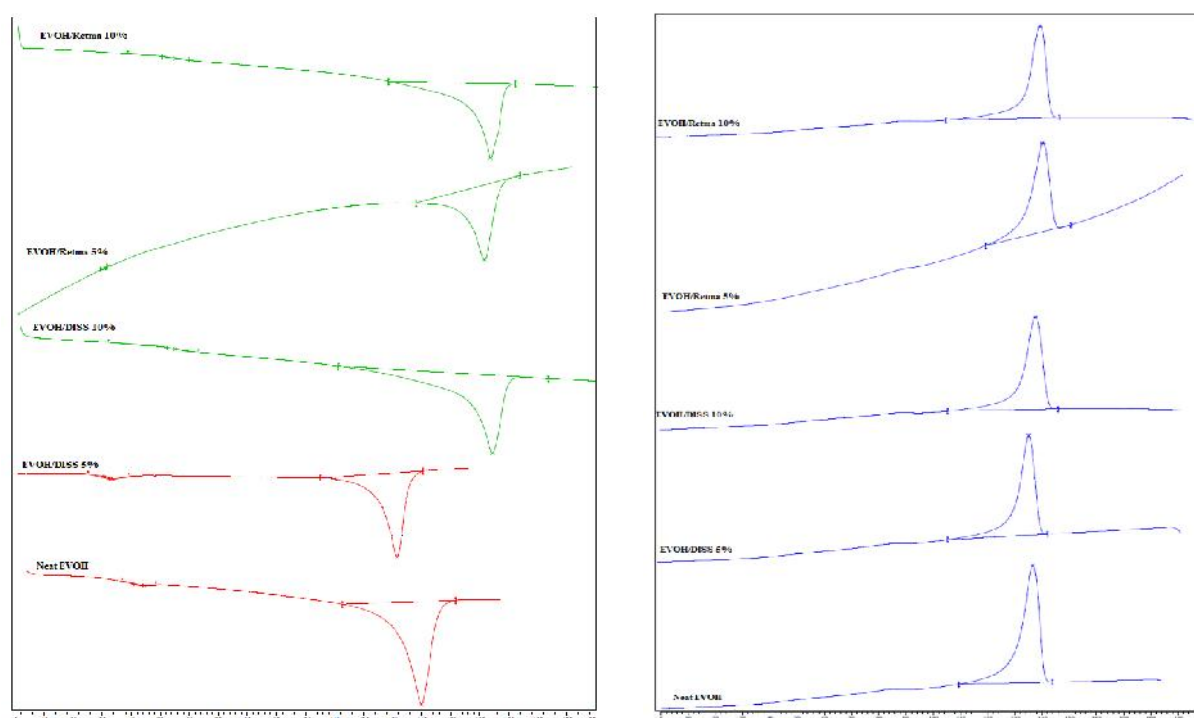
**Fig.6.** FTIR of EVOH/CF<sub>D</sub> (a) and EVOH/CF<sub>R</sub> (b) composites at 5 and 10%

The FTIR spectra obtained for the two types of composites presented a similar characteristic band. Two characteristics bands of common functional groups to both cellulose and EVOH structure appears at 3600–3000, 3000–2800 and 2900  $\text{cm}^{-1}$  assigned to the OH, CH and  $\text{CH}_2$  stretching vibration respectively [21]. The presence of cellulose in the EVOH matrix induces a decrease in the intensity of the OH band suggesting to the fact that the EVOH as well as the cellulose are polar and these hydroxyl groups present in the two structures tend to form inter and intra molecular hydrogen bonds [36,37], in the other hand, the intensity of OH increased when 10% of El Diss cellulose was incorporated suggesting an important number of OH groups

and are more important with El Diss fibers unlike El Retma fibers confirming the FTIR results of the fibers that the treatment undergone did not lead to complete elimination of lignin and hemicellulose and led to minimal OH exposure.

### b. Differential scanning calorimetry DSC

Thermal properties of EVOH/CF composites films were carried out by DSC. Figure 7 shows the DSC thermograms of neat EVOH, EVOH/CF<sub>D</sub> and EVOH/CF<sub>R</sub> composites at 5 and 10% and table I showed the melting point, melting enthalpy, and the glass transition temperature of neat EVOH, and EVOH/CF composites films.



**Fig.7.** DSC thermograms of treated and untreated EVOH/CF at 5% and 10% composites films

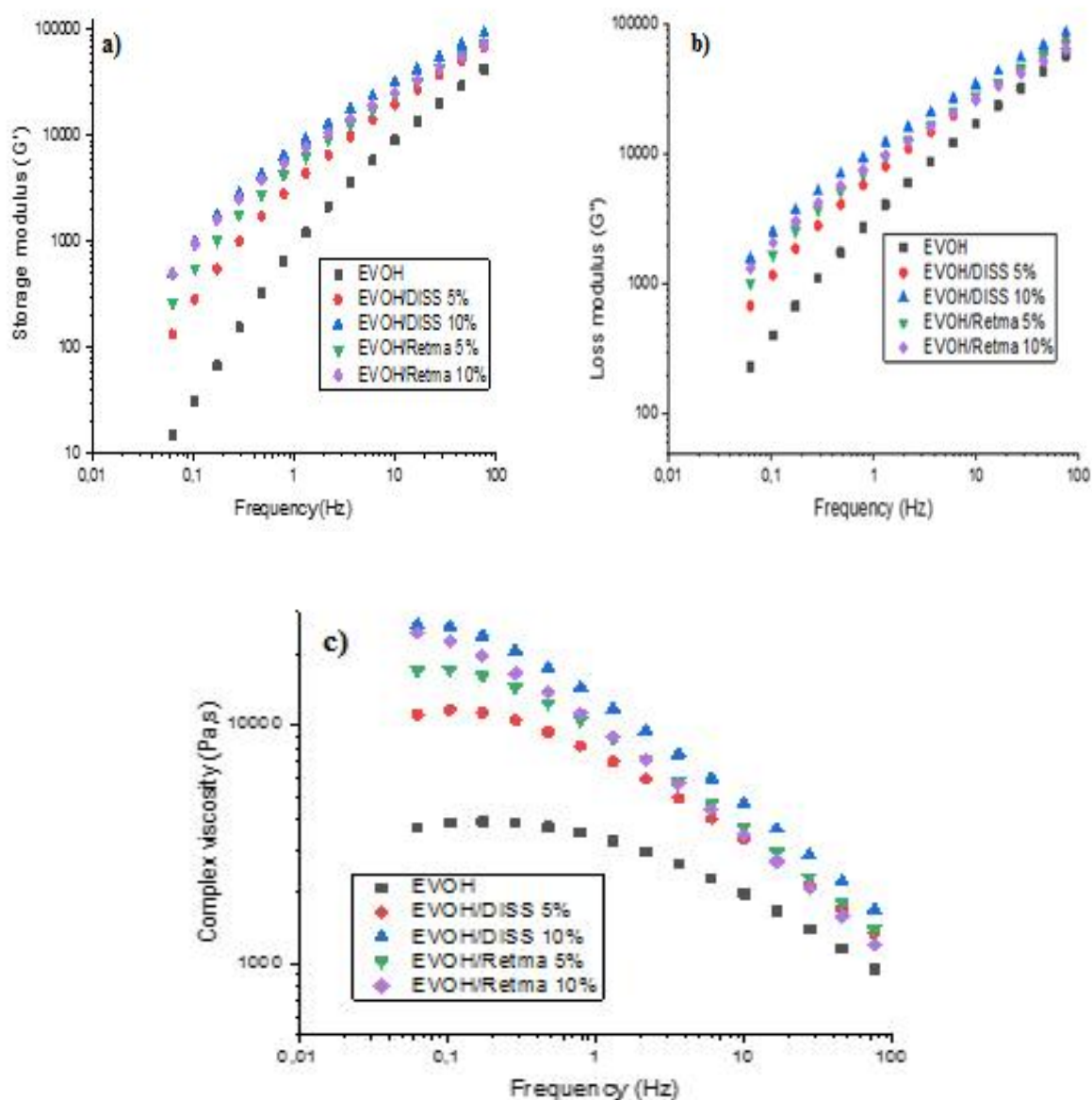
**Table 1.** The melting, cristallisation point and melting, cristallisation enthalpy of composites EVOH/CF

	T <sub>g</sub>	T <sub>c</sub>	H <sub>c</sub>	T <sub>m</sub>	H <sub>m</sub>
EVOH	71.90	142.33	41.64	167.83	-39.01
EVOH/CF <sub>D</sub> 5%	65.35	140.50	20.03	166.17	-18.90
EVOH/CF <sub>D</sub> 10%	67.27	140.33	49.83	165.50	-49.30
EVOH/CF <sub>R</sub> 5%	72.23	141.83	20.86	170.83	-20.50
EVOH/CF <sub>R</sub> 10%	50.67	141.50	45.95	163.00	-43.98

Compared to neat EVOH, the T<sub>g</sub> of EVOH / CF<sub>D</sub> at 5%, 10% and EVOH / CF<sub>R</sub> at 10% composites decreased from 71.9 to 65.35, 67.27 and 50.67 respectively, they also have lower (T<sub>c</sub>) and (T<sub>m</sub>) temperatures. The decrease in T<sub>m</sub> and especially the T<sub>g</sub> observed could be explain by the plasticization effect of cellulose on the EVOH as it was observed in the case of EVOH / SBO [22]. Whereas EVOH / CF<sub>R</sub> at 5% shows a slight higher (T<sub>g</sub>) and (T<sub>m</sub>) temperatures (72.23, 170.83 ° C respectively). The cristallisation enthalpy also reached to a lower temperatures (from 40 to around 20) for both composites with 5% of CF but increased slightly for 10% of fillers from 39.01 to 49.3 and 43.98 for CF<sub>D</sub> and CF<sub>R</sub> respectively. According to Dikobe and Luyt [23], Ethylene Vinyl Acetate Copolymer–Wood Fiber Composites has a decrease in H<sub>m</sub> as more WF was present in the samples.

### c. Rheological characterization

The elasticity modulus G', the elasticity complex modulus G'', and the complex viscosity of the EVOH composites with the different charge levels were represented in Figure 8.

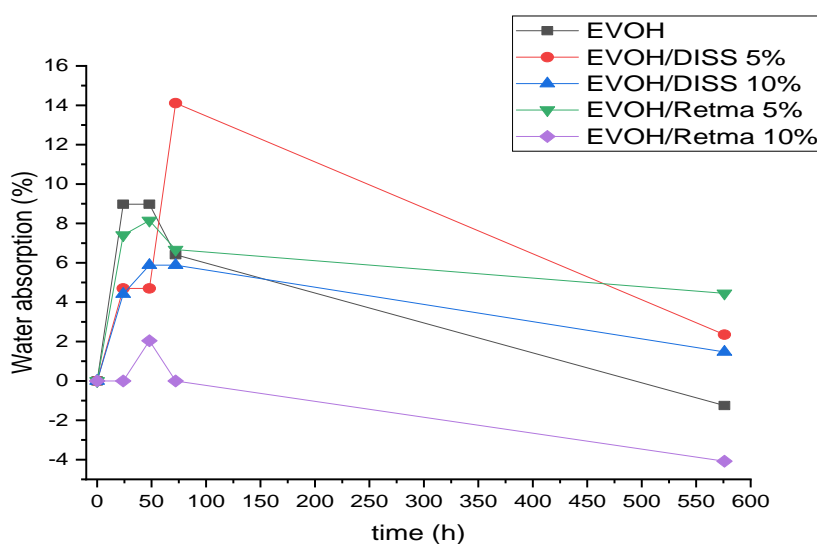


**Fig.8.** Spectra of  $G'$  (a),  $G''$  (b) modulus, complex viscosity (c)

The parameters values obtained for the EVOH composites were close to the value of the neat EVOH. The  $G'$  and  $G''$  modulus of EVOH/ $CF_D$  increased slightly and proportionally with fibers ratio, this results can be explain that EVOH is a hydrophile polymer which potentially make it compatible with the cellulose and it can also attributed to the surface fibers treatment where many hydroxyl groups are more exposed on the surface of the fibers which leads to the creation of bonds between the fibers and the matrix as showed in the FTIR results [38,20]. According to the study reported earlier about EVOH reinforced with water-soluble

lignin-like biopolymers (SLP) which found have similar or higher mechanical strength as the pristine synthetic polymer [39]. The composites HDPE/hard wood, HDPE/Flax, HDPE/ rice hulls, PP/ bleached hemp fiber, PP/Kenaf and PP/ white jute presented also an increasing in the Young's modulus with the fiber ratio. In the other hand, some researches found a contrary behaviour with increasing fiber ratio. This could be attributed to many factors like the incompatibility between the two phases, improper preparation process, fiber degradation or it can be also attributed to a poor dispersion [40]. The main objective of the rheological study is to better understand how the viscosity of these composites evolves. EVOH behavior is characteristic of pseudo-plastic materials, characterized by a Newtonian plateau at low frequencies and a decrease in viscosity as the frequency increases. Fibers effect on the composites viscosity increases is more obvious at lower frequency. This viscous behavior is determined by the nature of the interactions between the fibers, which can be easily explained in terms of the flow hindrance produced by the fibers [41]. For lower shear rates, the fibers are distributed with more disorder, which causes more fiber–fiber contact, further leading to increased viscosity. Adding the fiber to the polymer disrupts normal flow and hinders the mobility of the polymer segments, which increases the viscosity of the polymer [42].

#### d. Water absorption



**Fig.9.**Water absorption of EVOH/CF composites films at 5 and 10%

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The hydrophobic nature of EVOH leads to absorption of significant amounts of moisture in high relative humidity conditions [43], the polar hydroxyl group in EVOH structure cause its moisture sensitivity because they are not totally self associated, but partly isolated in the matrix which interacts with water molecules by hydrogen bonding [44,45]. The absorption of water in EVOH's composites occurs in the EVOH matrix through polar sites present in the polymer which increasing with ethylene content, the fibers where the -OH of the glucose is a potential site for water-absorption and the polymer -fibers interface [45-47]. The EVOH matrix has a mass gain of 8.97% which is compatible with the results reported earlier by others studies (Lagaron et al. [48] (2001) 9.0%, Aucejo et al. [49] (1999) 13%, Zhang et al. [50] (1999) 8.4% for EVOH films with 32 mol% ethylene). The water permeability decrease with increasing the degree of crystallinity, which increase with vinyl alcohol content and restricted chain mobility by intermolecular hydrogen-bonding interactions, but in the other hand, an increasing in vinyl alcohol unit means increasing hydroxyl groups that will interact with water molecule [45]. It is observed that the water absorption of the different composites is lower than that of the neat matrix. The water uptake of composite films filled with 5% decreased compared to the neat EVOH up to 4% and 1% for EVOH/CF<sub>R</sub> and EVOH/CF<sub>D</sub> respectively, which indicates that the water resistance capacity of EVOH is greatly enhanced by incorporation of the cellulosic fibers. It is also observed that composites with 10% cellulose are far less absorbent of water. An earlier study with MCS/PVA film, where the amount of starch in the film did not noticeably affect the water absorption capacity of the samples, this phenomenon can be explained that, the water in the nanopores may form hydrogen bonds with the polymer, blocking the nanopores and reducing water uptake [51]. In this case, the decreasing in water absorption observed with composites films is related to the fact that the polar groups contained in the structure of the fibers and matrix reacted together, this results are compatible with the FTIR results which indicates a decrease in OH number. Also, the water absorption decreased less in the case of CF<sub>R</sub>, thus confirming that the treatment of El Retma fibers could not expose more OH groups leading to a fewer interactions with the matrix and leaving more OH groups in the EVOH structure to interact with water. In the other hand, it's found that both the initial rate of water absorption and the maximum water uptake increases for all hemp fiber reinforced unsaturated polyester



composites samples as the fiber volume fraction increases. The author attributed the water absorption to the hemp fiber [52]. Similar experiments using a softwood kraft fiber as reinforcement in a xylan film was found that with 5% pulp fibers, specific water vapor transmission rate increases which increases in adding 10% pulp fibers to xylan [53]. The mass of the samples decreases as the immersion time increase, due to the partial solubilization of the PVA. The loss of mass is much more significant for the virgin matrix and decreases after the addition of the cellulosic fibers which is due to the interactions of OH groups contained on the cellulose surface which react with the OH in the PVA structure preventing it from solubilization.

#### 4. CONCLUSION

In this present study, El Diss and El Retma fibers are two new abundant sources which were treated to eliminate extracellular components (lignin and hemicellulose) and then incorporated to EVOH copolymer to obtain EVOH/CF composites. SEM images showed a significant morphological changes and FTIR results confirms the elimination of lignin and hemicellulose from the raw fibers. Thermal characterization indicate that the treated fibers degraded at slightly higher temperatures than the raw fibers, also suggesting that El Retma celluloses are not completely isolated.

After incorporation of the treated fibers into EVOH matrix, the FTIR results reported a formation of inter and intra molecular hydrogen bonds between EVOH matrix and cellulosic fibers. The  $G'$  and  $G''$  modulus has been improved after treated fibers addition in EVOH matrix which is compatible with literature reports and water absorptions results which decreased in the composites films because of the total involvement of the OH groups of the EVOH structure on the creation of inter and intramolecular bonds with cellulosic fibers preventing the formation of bonds with  $H_2O$ .

#### 5. ACKNOWLEDGEMENTS

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